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Bis[2,2'-ethylenedioxydibenzaldehyde bis(thiocarbohydrazide)] tris(pyridine) solvate¹

Suchada Chantrapromma,^a* Ibrahim Abdul Razak,^b Hoong-Kun Fun,^b Chatchanok Karalai,^a Hao Zhang,^c Fu-Xin Xie,^c Yu-Peng Tian,^c Wen Ma,^c Yin-Han Zhang^d and Shi-Sheng Ni^c

^aDepartment of Chemistry, Faculty of Science, Prince of Songkla University, Hat-Yai, Songkhla 90112, Thailand, ^bX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, ^cDepartment of Chemistry, Anhui University, Hefei, Anhui 230039, People's Republic of China, and ^dDepartment of Chemistry, University of Science and Technology of China, Hefei, Anhui 230026, People's Republic of China

Correspondence e-mail: suchada@ratree.psu.ac.th

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The title compound, a novel 30-membered 2:2 macrocyclic thiocarbohydrazone, $C_{34}H_{32}N_8O_4S_2\cdot 3C_5H_5N$, has been prepared and crystallographically characterized. The molecule of the compound is twisted. One dioxabutane group is boat-like in shape, whereas the other is highly disordered. The crystal structure is stabilized by inter- and intramolecular hydrogen bonds.

Comment

Thiocarbohydrazide, H₂NHNC(S)NHNH₂, and its Schiff base derivatives are interesting ligand systems because of the availability of several different kinds of potential donor sites (Bustos *et al.*, 1990; Bacchi *et al.*, 1996; Souza *et al.*, 1994). Recently, many open-chain ligands of thiocarbohydrazone have been rapidly developed and studied in detail. The aim of the present research is the design of a new macrocyclic ligand involving mixed N-, S- and O-donor atoms, and we describe here the synthesis and characterization of a novel 30-membered 2:2 macrocyclic thiocarbohydrazone, (I), obtained by the condensation of thiocarbohydrazide with 1,4-bis(2-formylphenyl)-1,4-dioxabutane.

As shown in Fig. 1, the molecule of (I) is twisted. The S1– C1 [1.666 (3) Å] and S2–C18 [1.657 (3) Å] distances indicate that they are double bonds (Bustos *et al.*, 1990; Bacchi *et al.*, 1996). This suggests the presence of a thioketo form in the solid state. The torsion angle along the O1–C9–C10–O2 dioxabutane moiety is 64.8 (4)°. This dioxabutane group is boat-like in shape, whereas the other is highly disordered (atoms O4, C26 and C27 are disordered). Each unit of the macrocyclic ring, consisting of the thiocarbohydrazone and the axially substituted phenyl rings, adopts a planar conformation. The dihedral angles are $15.4 (2)^{\circ}$ between rings C3–C8 and C28–C33, and $12.6 (2)^{\circ}$ between rings C11–C16 and C20–C25. The twist is introduced by the dioxabutane bridges connecting the two units. The dihedral angle between the two thiocarbohydrazone groups is $30.98 (8)^{\circ}$.



The three molecules of pyridine present as solvate in (I) help to stabilize the structure by forming intra- and intermolecular hydrogen bonds with the macrocyclic ring; details are given in Table 2. The C-H···Cg(π -ring) interactions [Cg1 = C20-C25, Cg2 = C28-C33, Cg3 = N10 and C40-C44 (one of the three pyridine molecules), and Cg4 = C11-C16], *i.e.* the interaction between a phenyl H atom and a centre of gravity of a pyridine ring, and between the H atoms of the pyridine rings and the centres of gravity of the phenyl rings, play a major role in the packing and molecular arrangement in the crystal (Table 2).



Figure 1

The molecular structure of (I) showing 50% probability displacement ellipsoids and the atom-numbering scheme. H atoms and the pyridine solvate molecules have been omitted for clarity.

Experimental

A solution of 1,4-bis(2-formylphenyl)-1,4-dioxabutane (10 mmol) in ethanol (150 ml) was dripped slowly into a refluxing solution of thiocarbohydrazide (10 mmol) in aqueous ethanol (100 ml), followed by the addition of concentrated hydrochloric acid (0.25 ml). The reaction mixture was refluxed for 10 h with stirring and cooled to room temperature. The yellow precipitated powder of (I) was filtered and washed thoroughly with water and ethanol, and then air dried. A crystal suitable for X-ray diffraction was obtained by evaporation from a benzene–pyridine (1:1) solution.

¹ Systematic name: 2,5,25,28-tetraoxa-13,14,16,17,36,37,39,40-octaazapenta-cyclo[$40.4.0.0^{6,11}.0^{19,24}.0^{29,34}$]hexatetraconta-6,8,10,12,17,19,21,23,29,31,33,35,-40,42,44,46-hexadecaene-15,38-dithione tris(pyridine) solvate.

Crystal data

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$C_{34}H_{32}N_8O_4S_2{\cdot}3C_5H_5N$	$D_x = 1.278 \text{ Mg m}^{-3}$
$M_r = 918.10$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/c$	Cell parameters from 5918
a = 12.1752 (2) Å	reflections
b = 22.2029 (1) Å	$\theta = 2.77 - 33.11^{\circ}$
c = 19.8479 (4) Å	$\mu = 0.168 \text{ mm}^{-1}$
$\beta = 117.168 (1)^{\circ}$	T = 293 (2) K
$V = 4773.42 (13) \text{ Å}^3$	Block, yellow
Z = 4	$0.37 \times 0.33 \times 0.28 \text{ mm}$
Data collection	
Siemens SMART CCD area-	10 907 independent reflections
detector diffractometer	5918 reflections with $I > 2\sigma(I)$
ω scans	$R_{\rm int} = 0.043$
Absorption correction: empirical	$\theta_{\rm max} = 27.5^{\circ}$
(SADABS; Sheldrick, 1996)	$h = -14 \rightarrow 15$
$T_{\min} = 0.941, T_{\max} = 0.955$	$k = -28 \rightarrow 23$
30 445 measured reflections	$l = -25 \rightarrow 23$
Refinement	
Refinement on F^2	$w = 1/[\sigma^2(F_0^2) + (0.0460P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.068$	+ 2.8579P]
$wR(F^2) = 0.166$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.051	$(\Delta/\sigma)_{\rm max} < 0.001$
10 907 reflections	$\Delta \rho_{\rm max} = 0.46 \ {\rm e} \ {\rm \AA}^{-3}$
593 parameters	$\Delta \rho_{\rm min} = -0.49 {\rm e} {\rm \AA}^{-3}$
H-atom parameters constrained	

Table 1

Selected geometric parameters (Å, °).

\$1-C1	1.666 (3)	S2-C18	1.657 (3)
N1-C1-N3 N1-C1-S1 N3-C1-S1	114.6 (2) 119.5 (2) 125.9 (2)	N6-C18-N7 N6-C18-S2 N7-C18-S2	113.2 (2) 126.3 (2) 120.5 (2)
01-C9-C10-O2	64.8 (4)		

After checking their presence in the difference map, all H atoms were fixed geometrically and allowed to ride on the atoms to which they were attached, with $U_{\rm iso} = 1.2U_{\rm eq}$ of the parent atoms, and with C-H = 0.93-0.97 Å and N-H = 0.86 Å. The ratio of the major to the minor component of the disordered dioxabutane group is 0.56:0.44. The bond lengths of the disordered group were restrained to be the same as those of the non-disordered dioxabutane group.

Data collection: *SMART* (Siemens, 1996); cell refinement: *SAINT* (Siemens, 1996); data reduction: *SAINT*; program(s) used to solve

Table 2

Hydrogen-bonding geometry (Å, $^{\circ}$).

Cg1 is the centroid of the C20–C25 ring, Cg2 is the centroid of the C28–C33 ring, Cg3 is the centroid of the N10/C40–C44 ring (pyridine solvate) and Cg4 is the centroid of the C11–C16 ring.

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdot \cdot \cdot A$	$D - H \cdots A$
$N1 - H1A \cdots N9^{i}$	0.86	2.135	2.990 (4)	173
$N3-H3A\cdots N10$	0.86	2.321	3.061 (4)	144
$N7 - H7A \cdots N11$	0.86	2.173	3.008 (4)	164
$C40-H40A\cdots Cg1$	0.93	2.757	3.453 (5)	132
$C45-H45A\cdots Cg2$	0.93	2.868	3.673 (5)	146
$C13 - H13A \cdots Cg3^{ii}$	0.93	2.752	3.522 (4)	141
$C36-H36A\cdots Cg2^{iii}$	0.93	2.942	3.746 (5)	146
C46-H46 A ··· Cg 4 ^{iv}	0.93	2.941	3.769 (5)	149
Symmetry codes: (i) x	-1. v. z: (ii)	$-x$, $y - \frac{1}{2}, \frac{1}{2} -$	<i>z</i> : (iii) $1 - x$, <i>y</i>	$-\frac{1}{2},\frac{1}{2}-7$; (iv)

 $1 - x, \frac{1}{2} + y, \frac{1}{2} - z.$

structure: *SHELXTL* (Sheldrick, 1997); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 1990).

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1435). Services for accessing these data are described at the back of the journal.

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